

Study of the Activated GaAs Surface for Application as an Electron Source in Particle Accelerators

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Outline

- Introduction
- Experimental setup
- Preliminary results
- Summary

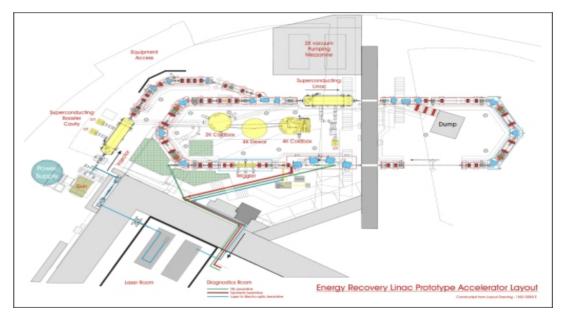




Introduction

• The Accelerators and Lasers in Combined Experimental (ALICE) is a 35 *MeV* energy recovery test facility currently under construction at Daresbury Laboratory









Photocathode for ALICE

- The type III-V semiconductor, especially GaAs, photocathode has been focused.
- This type of photocathode has high quantum efficiency and generates high polarised electron.
- However, it need to be activated to the negative electron affinity (NEA) state.
- This photocathode need to be kept in the extreme high vacuum (XHV) conditions.



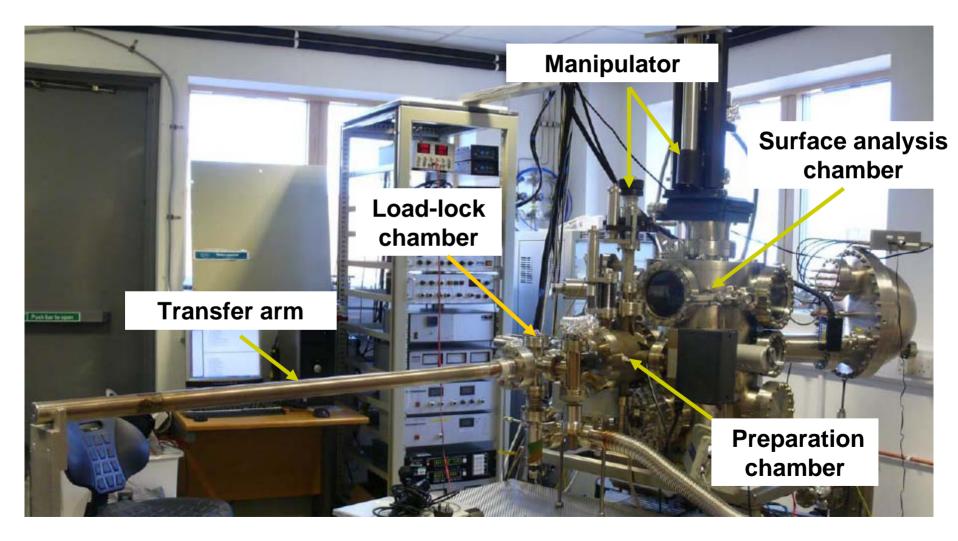
Photocathode R&D at Daresbury Laboratory

- Study a range of photocathode material in detail, investigating and developing the various fundamental processes, such as preparation process to maximise their performance.
- The initial focus will be on GaAs, but alternative materials such as Cs₂Te and InGaP will be also studied.





Experimental Setup

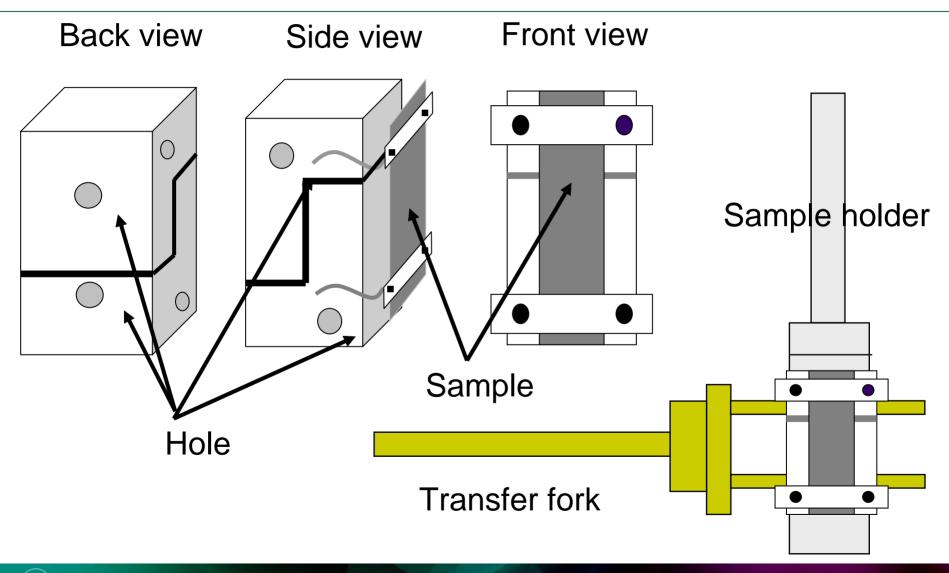






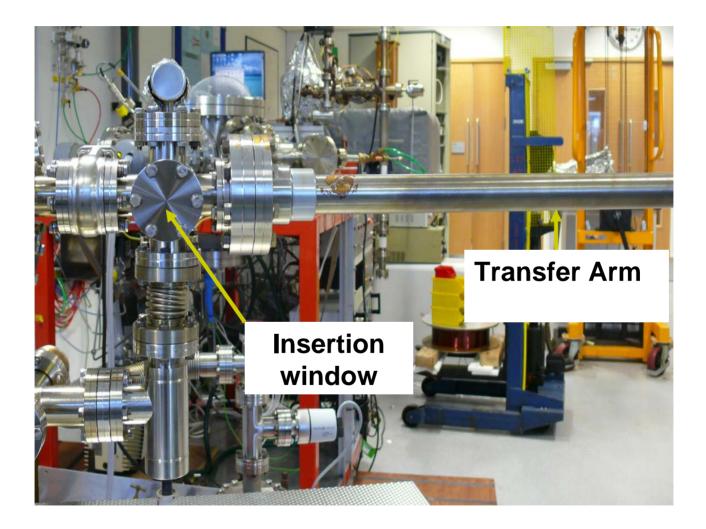
Sample mounted

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Load-lock Chamber





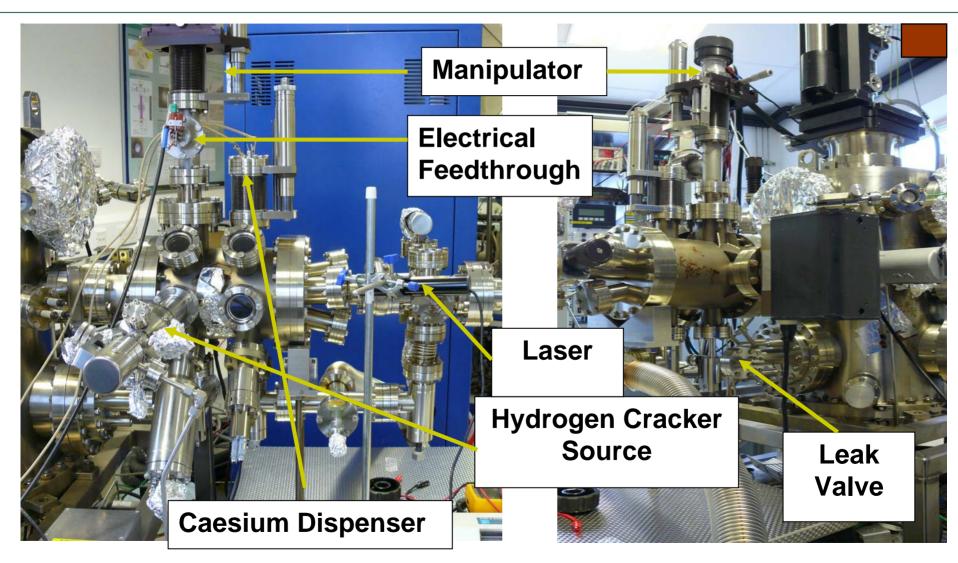


Preparation Process

- The GaAs sample need to be cleaned before activated to the NEA state.
- Choices of cleaning process
 - Heat in the vacuum
 - Atomic hydrogen cleaning
- Activated to the NEA state by applying caesium and oxidant (either O₂ or NF₃) onto its clean crystal surface.
- The standard Yo-Yo method is employed at room temperature.



Preparation Chamber

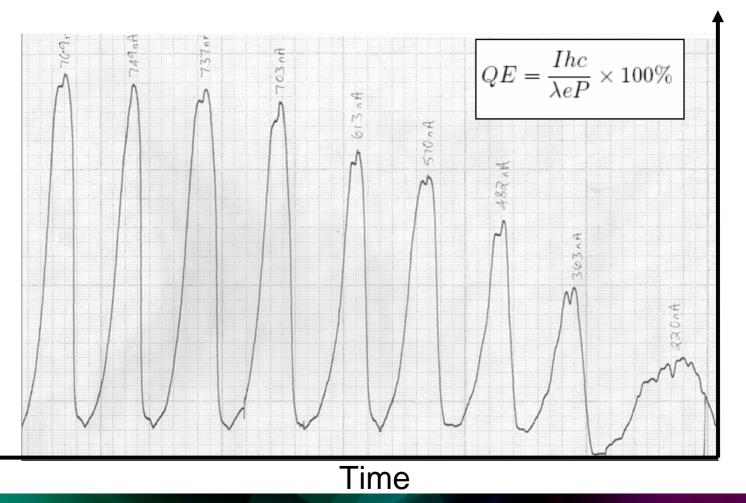




Photocurrent



The Yo-Yo method



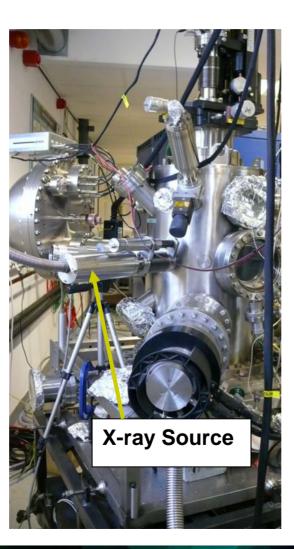


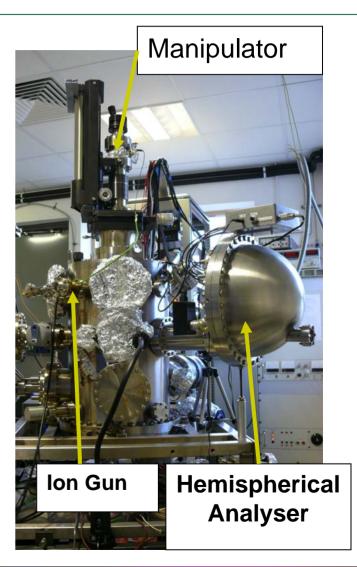


Surface Analysis Chamber



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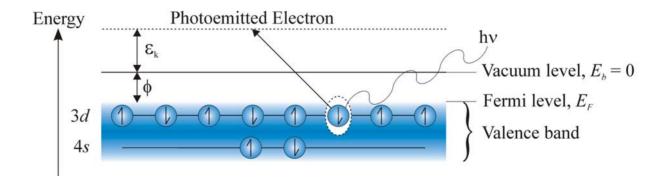






X-ray Photoelectron Spectroscopy (XPS) Technique

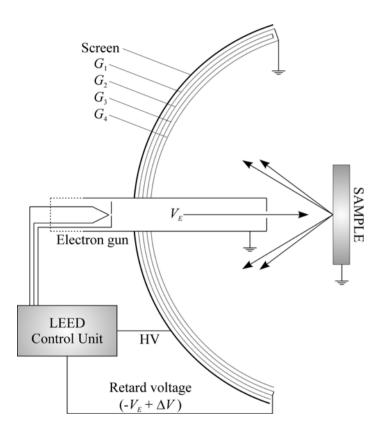
- Used to investigate the chemical composition of a material surface.
- Irradiates a material surface with soft x-rays, $Mg K\alpha$ or $AI K\alpha$ in a UHV.
- Cause the electrons to be emitted by the photoelectric effect.
- The chemical composition of a material is identified by analysing the kinetic energy and the number of emitted electrons.







Low Energy Electron Diffraction (LEED) Technique



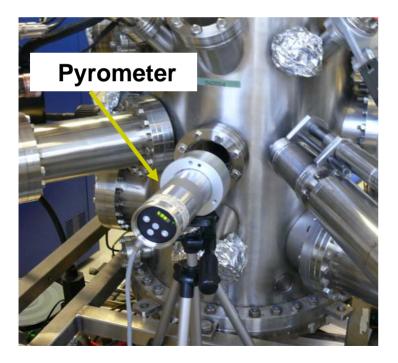
- Used to determine the surface structures of a material
- Incidents a beam of low energy electrons into the sample surface
- The electron beam will be diffracted following the Bragg condition.
- Determine the surface structure from the intensity and diffraction pattern of the back-scattered electrons

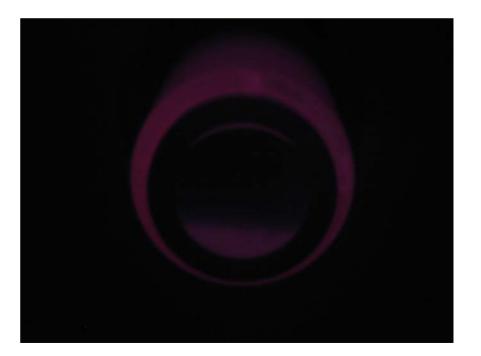




Effect of the temperature in heat cleaning

- Heat for 1 hour in the UHV
- Observe the temperature by using the pyrometer

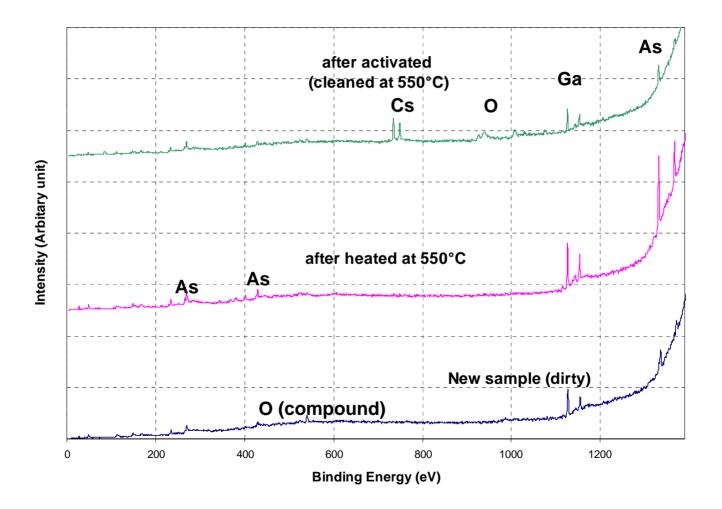








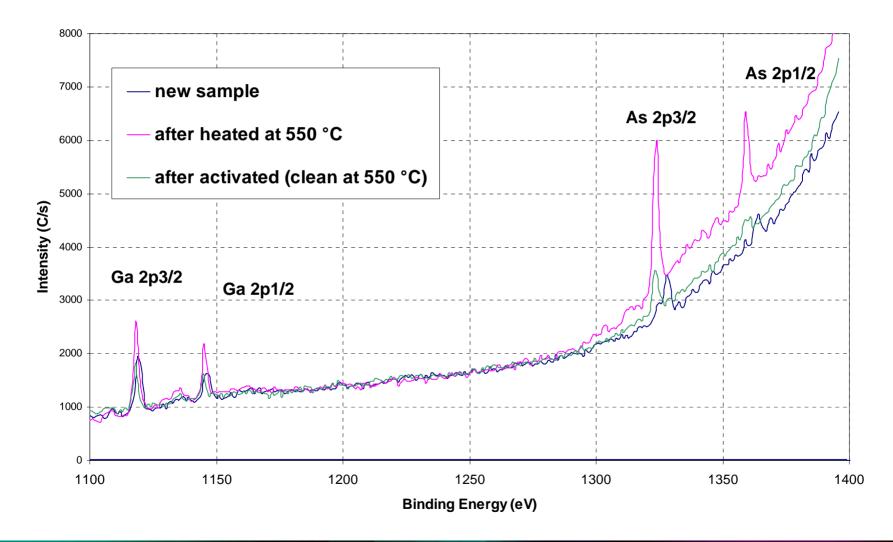
ASTeC







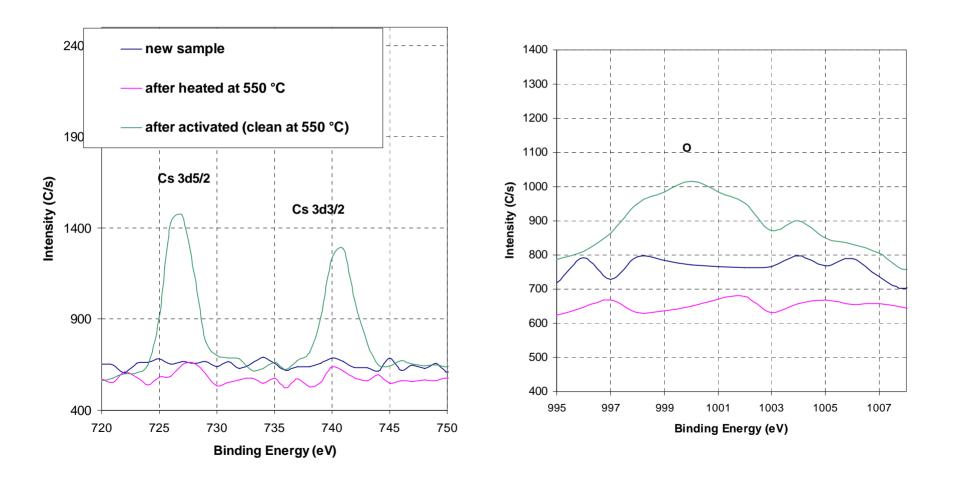
XPS results







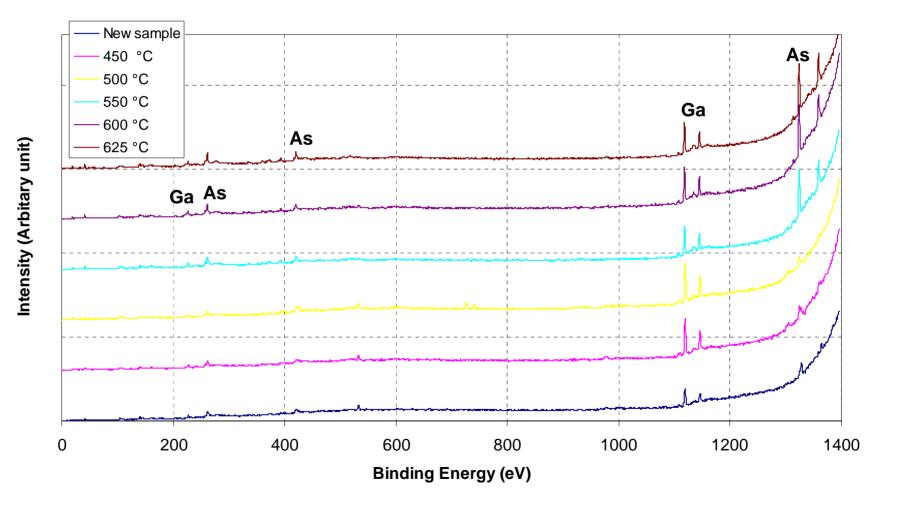
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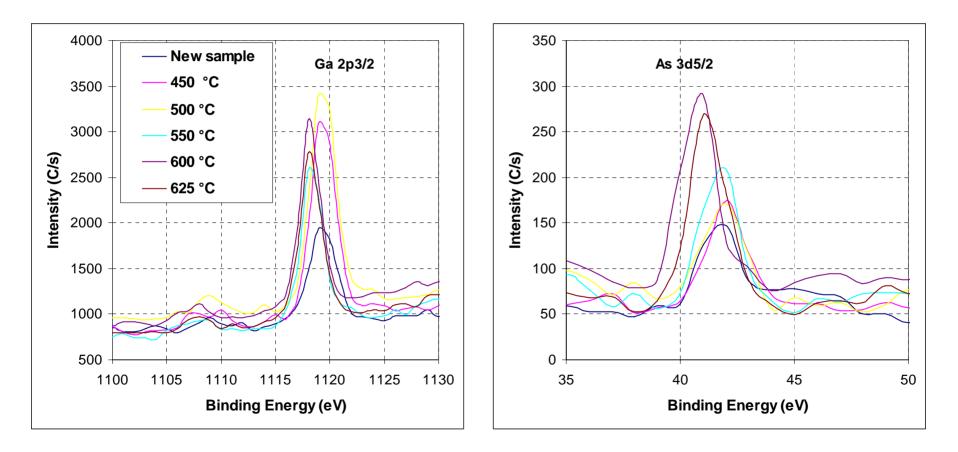
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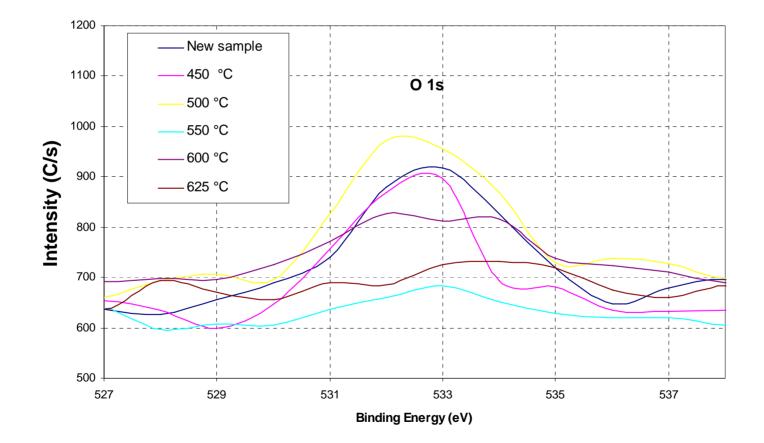
Heat Cleaning







ASTeC

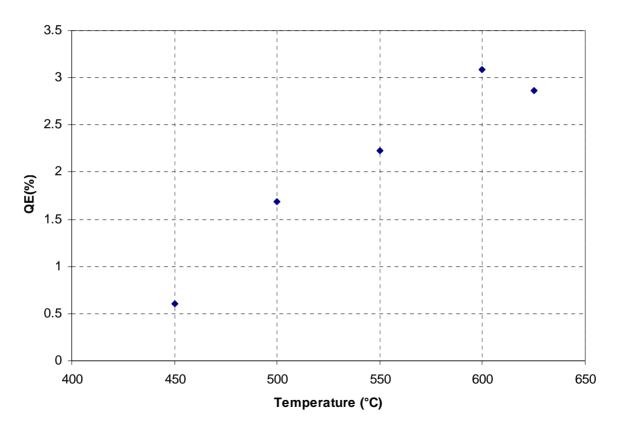






Quantum efficiency (QE)

The effect of the heat temperature on the QE of the GaAs photocathode

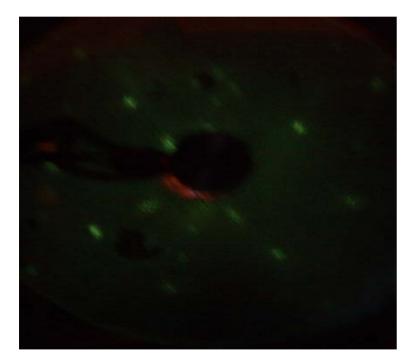






LEED Pattern

 Picture of LEED diffraction of the GaAs after heated at 600 C for 1 hour









Summary

- The vacuum system which supports the surface science techniques have been setup to study the activation process of the NEA GaAs photocathode.
- The effect of the temperature in heat cleaning process is studied by using the XPS and LEED.
- We have presently just finished the installation and the testing of components in our test chamber. For the future works, the atomic hydrogen cleaning and lifetime will be studied.





• Thank you for your attention



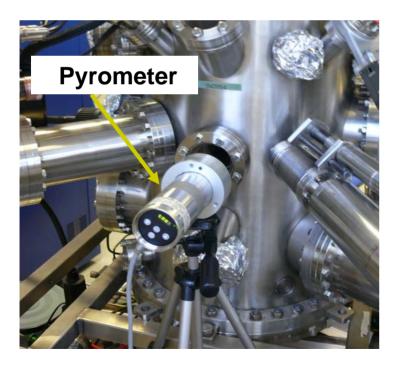


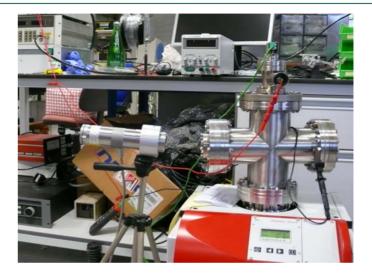


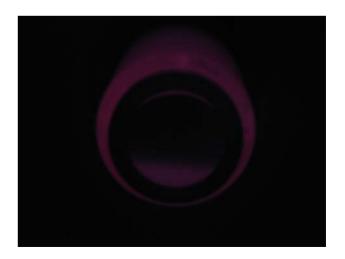


Effect of the temperature in heat cleaning

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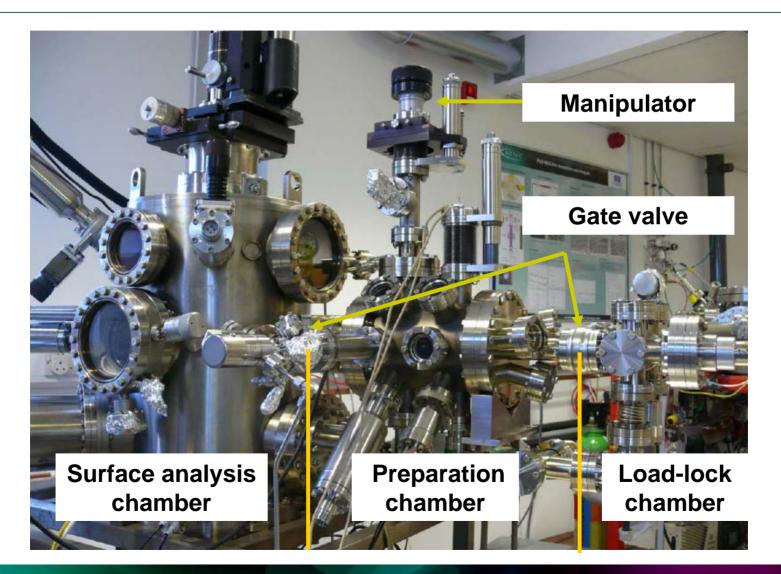








Experimental Setup

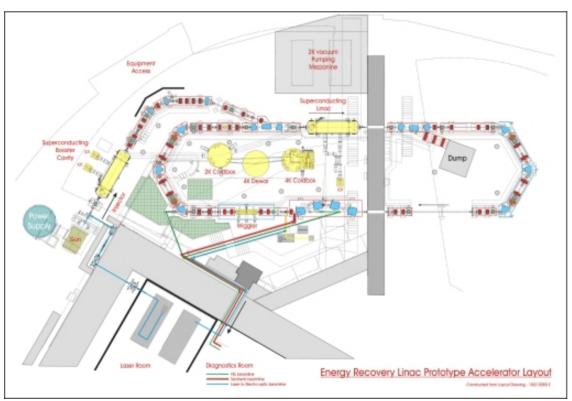






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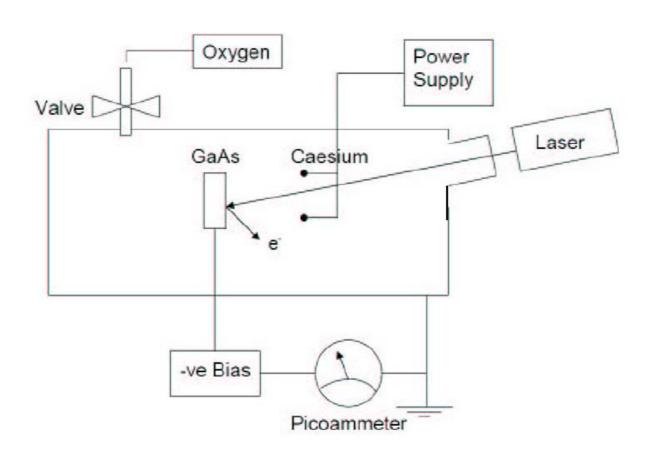


- The photoinjector is based on the JLab FEL DC photocathode gun.
- A CW current of 6.5 *m*A as a pre-cursor to the 100 *m*A
- Normalised emittance of < 1 π mm mrad.





Diagram of the activation process







Low Energy Electron Diffraction (LEED) Technique

